



MAX PLANCK INSTITUTE FOR DYNAMICS OF COMPLEX TECHNICAL SYSTEMS MAGDEBURG

TU Clausthal

Symposium on Insights into Gas Diffusion Electrodes

From Fundamentals to Industrial Applications





General Information

Conference Venue

Veranstaltungszentrum Festung Mark Hohepfortewall 1 39104 Magdeburg

Internet Access

WiFi access is available for free throughout the conference venue. As WiFi can be used by all participants, a loss of efficiency is possible. Network: Zukunftswerkstatt Password: ZKW_2017

Book of Abstracts

Book of Abstracts (lectures and posters) is available online for participants of the meeting "Insights into Gas Diffusion Electrodes: From Fundamentals to Industrial Applications": https://www.mpi-magdeburg.mpg.de/gde-symposium-2019

Social Programme

Tuesday, 24 September 2019, 19:00 Conference Dinner (Festung Mark)



Contact

Thomas Turek Phone: +49 5323 722 184 E-mail: turek@icvt.tu-clausthal.de Wednesday, 25 September 2019, 14:00 Guided Tour MPI Magdeburg



Fotos: © Festung Mark

Susanne Wandenälis Phone : +49 391 6110 317 E-mail : gde2019@mpi-magdeburg.mpg.de

General Information

Scientific Scope

The Symposium on "Insights into Gas Diffusion Electrodes: From Fundamentals to Industrial Applications" is an initiative of the Research Unit 2397 approved by German Research Foundation (DFG) in July 2016. The main research focus of the project "Multiscale Analysis of Complex Three-Phase Systems: Oxygen Reduction at Gas Diffusion Electrodes in Aqueous Electrolyte" are the complex processes within gas diffusion electrodes (GDEs). GDEs have broad applications in different electrochemical devices such as (bio)-fuel cells, (bio)-electrolyzers, and batteries for energy as well as inorganic and organic synthesis applications. Although very different chemistries and materials are involved, all of these examples share some similarities related to so-called "three phase boundaries" as well as pronounced

Plenary Lectures

Multiphase transport in gas diffusion electrodes for fuel cells and electrolyzers: Modelling, imaging, and design Aimy Bazylak, Mechanical & Industrial Engineering, University of Toronto, Canada

mass transfer resistances related to slow diffusivity and/or low solubility of gases.

Understanding and Optimizing Water Transport in Gas Diffusion Layers F. Büchi, Paul Scherrer Institut, Switzerland

Durability of carbon-supported nanostructured electrocatalysts in alkaline environments: more noble is not necessarily more stable Marian Chatenet, Grenoble Institute of Technology, France

How can gas-diffusion electrodes leverage enzymatic fuel cells performance Elisabeth Lojou, CNRS-Aix Marseille Université, France

Scientific Committee

Ulrike Krewer Ingo Manke Ulrich Nieken Christina Roth Wolfgang Schuhmann Thomas Turek Tanja Vidaković-Koch TU Braunschweig Helmholtz Zentrum Berlin University Stuttgart University Bayreuth RBU Bochum TU Clausthal MPI Magdeburg

Sponsors

Sponsors

Forschungsgemeinschaft

German Research Foundation (Deutsche Forschungsgemeinschaft, DFG) is the self-governing organisation for science and research in Germany. It serves all branches of science and the humanities. In organisational terms, the DFG is an association under private law. Its membership consists of German research universities, non-university research institutions, scientific associations and the Academies of Science and the Humanities. DFG supports this symposium within the framework of the research unit FOR 2397. www.dfg.de

www.for2397.tu-clausthal.de



The International Society of Electrochemistry (ISE) was founded in 1949 by leading European and American Electrochemists to serve the growing needs of electrochemistry in becoming a modern scientific discipline. Since then the association has evolved and now comprises about 3000 individual members and more than 20 Corporate Members (teaching institutions, non-profit-making research organizations and learned societies) and Corporate Sustaining Members (industrial and commercial organizations). Its membership comes from more than 70 countries and is organized in over 40 regional sections. ISE is a non-profit-making organization with its seat in Lausanne, Switzerland. www.ise-online.org



Otto von Guericke University Magdeburg has a distinctive profile. It aims to create a lean and sharply-defined structure with its main focus of expertise in the traditional areas of engineering, the natural sciences and medicine. It also views economics and management and the social sciences and humanities as essential disciplines for a modern university in the information age. www.uni-magdeburg.de/en







The Gesellschaft Deutscher Chemiker (GDCh) is the largest chemical society in continental Europe with members from academe, education, industry and other areas. The GDCh supports chemistry in teaching, research and application and promotes the understanding of chemistry in the public. The Electrochemistry division of GDCh was founded in 1960. The increasing number of members (today, about 450) reflects the resonance in all areas of science and engineering. The division's established aim is to comprehensively promote electrochemistry from the fundamentals up to applications. http://en.qdch.de

Covestro is a world's leading producer of high-quality chemistry based materials. The company's products and application solutions provide the basis for key industries and are ubiguitous in modern life. The focus is on innovative, sustainable solutions for the big challenges of our time - such as climate change, increasing scarcity of resources, growing mobility and urbanization. Covestro has some 16,800 employees across the globe and generated sales of 14.6 billion euros in 2018.

www.covestro.com/en

C3 Prozess- und Analysentechnik GmbH, founded in 1986, represents highly respected international manufacturers of analytical and laboratory instrumentation in Germany, Austria, and Switzerland.

With the central office near Munich and offices around Germany we provide application consulting, sales, installation and training as well as after sales support for our customers. Our portfolio related to battery research, development and testing covers potentiostats and EIS measurement systems from Gamry Instruments (USA) and electrochemical accessories from ALS (Japan). For battery cells and packs, as well as material and safety testing, we offer isothermal and adiabatic calorimeters from tht (UK) and for mixing electrode pastes we can provide Thinky planetary mixers (Japan).

www.c3-analysentechnik.de

Conference Program

Monday, 23 September 2019

12:00	Registration
13:00	Welcome
Chair	Ulrich Nieken
13:10	PLENARY LECTURE Understanding and Optimizing Water Transport in Gas Diffusion Layers F. Büchi, Paul Scherrer Institut
14:00 O-01	FIB Tomography of silver-based gas diffusion electrodes with different silver-PTFE ratios, M. Paulisch, Helmholtz Zentrum Berlin
14:20 O-02	Prediction of Electrolyte Distribution in Technical Gas Diffusion Electrodes: From Imaging to SPH Simulations, T. Mager, University of Stuttgart
14:35 O-03	3D pore-scale lattice Boltzmann modelling of porous electrodes in fuel cells and batteries, Cai Qiong, University of Surrey
14:55 O-04	Estimation of effective parameters from pore network simulations N. Vorhauer, Otto von Guericke University Magdeburg
15:15	Coffee Break
Chair	Tanja Vidaković-Koch
15:45	INVITED LECTURE Optimizing Local Reaction Conditions in Porous Electrodes: Insights from Theory and Computation, M. Eikerling, Forschungszentrum Juelich
16:10 O-05	Models and experiments for the analysis of ORR product accumulation with- in porous oxygen gas diffusion electrodes, S. Eggermont, KU Leuven
16:25 O-06	Dynamics and States of Oxygen Depolarized Cathodes: A Model-Based Analysis, M. Röhe, TU Braunschweig
16:40 O-07	Spatially resolved model of oxygen reduction reaction in porous gas- diffusion Electrodes, D. Franzen, TU Clausthal
16:55 O-08	Dynamic Changes of Cathode Morphology During Operation of $Li-O_2$ Batteries: A Model-Based Analysis, S. Kynast, TU Braunschweig
17:10	Short Introduction of Poster Programm
17:15 – 19:15	Poster Party (with snacks)

Conference Program

Tuesday, 24 September 2019

Chair	Wolfgang Schuhmann
09:00	PLENARY LECTURE How can gas-diffusion electrodes leverage enzymatic fuel cells perfor- mance, E. Lojou, CNRS-Aix Marseille Université
9:50 O-09	Polymer Modified Gas Diffusion Electrodes Containing Hydrogenases or Their Artificial Mimics as Active H ₂ Oxidation Catalysts J. Szczesny, Ruhr-Universität Bochum
10:05 O-10	Development and upscaling of gas diffusion electrodes for environmental applications in wastewater treatment and electrosynthesis of chemicals, D. Pant, Flemish Institute for Technological Research (VITO)
10:25	Coffee Break
Chair	Jürgen Kintrup
10:55	INVITED LECTURE CO ₂ electroreduction. From catalyst design to gas diffusion electrodes C. Andronescu, Universität Duisburg-Essen
11:20 O-11	Electrochemical conversion of CO ₂ to methanol – A feasible reaction for large scale?, S. Geiger, German Aerospace Center
11:40 O-12	The electrolyte matters: stable systems for high rate electrochemical CO ₂ reduction, S. Kriescher, RWTH Aachen University
12:00 O-13	Examination of gas-diffusion electrodes applied in CO₂ electrolysis employ- ing electrochemical impedance spectroscopy (EIS), F. Bienen, German Aerospace Center
12:15	INVITED LECTURE Benchmarking and tuning the oxygen reduction activity on Pt-based cata- lysts: from model studies to gas diffusion electrode measurements M. Escudero-Escribano, University of Copenhagen
12:40 O-14	Beyond Equivalent Circuit Approach: Computer-Enhanced Nonlinear Frequency Response Method L. Živković, Faculty of Technology and Metallurgy, University of Belgrade
12:55	Lunch Break

Conference Program

Tuesday, 24 September 2019

Chair	Thomas Turek
14:00	PLENARY LECTURE Durability of carbon-supported nanostructured electrocatalysts in alkaline environments: more noble is not necessarily more stable! Marian Chatenet, Grenoble-INP
14:50 O-15	Nanoscale Studies Solve Macroscopic Problems: De- and Reactivation of Oxygen Depolarized Cathodes Under Process Conditions D. Öhl, Ruhr-Universität Bochum
15:05 O-16	Probing the degradation pathways of PEM water electrolysers under dy- namic operating conditions with advanced microscopy methods G. Papakontantinou, MPI Magdeburg
15:25	Coffee Break and Poster Discussions
Chair	Christina Roth
16:25 O-17	Analysing the performance instability of H_2 - O_2 Alkaline Membrane Fuel Cell due to water imbalance, P. Khadke, Freie Universität Berlin
16:45 O-18	Novel nanoparticles synthesis method by gas-diffusion electrocrystalli- zation, G. Pozo, Flemish Institute for Technological Research (VITO)
17:05 O-19	Three-phase activity of unsupported high-surface area catalysts towards oxygen reduction in the gas diffusion electrode G. Sievers, Leibniz Institute for Plasma Science and Technology
17:20 O-20	Using GDE half-cell experiments to close the gap between fundamental and applied fuel cell electrocatalysis K. Ehelebe, Helmholtz-Institute Erlangen-Nürnberg
17:40 – 18:05	INVITED LECTURE Towards realistic testing conditions for electrocatalysts: Gas diffusion electrodes incorporated into half-cell setups, M. Arenz, University of Bern
19:00 – 22:00	Conference Dinner (Festung Mark)

Conference Program

Wednesday, 25 September 2019

Chair	Ingo Manke
09:00	PLENARY LECTURE Multiphase transport in gas diffusion electrodes for fuel cells and electroly- zers: Modelling, imaging, and design, Aimy Bazylak , University of Toronto
9:50 O-21	Acid Migration in the Gas Diffusion Electrode of High-Temperature Polymer Electrolyte Membrane Fuel Cells, R. Zeis, Karlsruhe Institute of Technology
10:10 O-22	Understanding the influence of different porous microstructures derived from diverse drying methods of the cathode catalyst layer in PEMFC K. Talukdar, German Aerospace Center, Stuttgart
10:25	Coffee Break
Chair	Ulrike Krewer
10:55	INVITED LECTURE Operando X-ray Computed Tomography Study of Water Management in Gas Diffusion Electrodes of PGM-free Polymer Electrolyte Fuel Cells I. Zenyuk, The Henry Samueli School of Engineering, University of California
11:20 O-23	A Ternary-based composite as cost-effective, efficient and durable catalyst layer for bifunctional gas diffusion electrodes, E. Marini, Zentrum für Son- nenenergie- und Wasserstoffforschung Baden-Württemberg (ZSW)
11:35 O-24	Development of new carbon free electrode for application in secondary zinc-air cells, G. Raikova, Bulgarian Academy of Sciences
11:55 O-25	Effect of wetting behaviour on the electrochemical performance of bifunc- tional gas diffusion electrodes for electrically rechargeable zinc-air batteries M. Fink, University of Bayreuth
12:10 O-26	Stable carbon materials for metal/air batteries and PEM fuel cells GDEs JF. Drillet, DECHEMA-Forschungsinstitut

Conference Program

Wednesday, 25 September 2019

12:30	INVITED LECTURE Gas Diffusion Electrodes in Industry - from Chlorine to CO ₂ Reduction J. Kintrup, Covestro Deutschland AG
12:55	Closing
14:00	Guided Tour MPI Magdeburg



Poster Program

	Batteries
P-01	The Developments of Carbon-based Gas Diffusion Electrodes (GDEs) at IEES B. Abrashev, Bulgarian Academy of Sciences
P-02	Controllable Preparation of Graphene-based air electrode for Efficient Rechargeable Mg-Air Batteries D. Fu, Research Institute of Tsinghua University in Shenzhen
P-03	Highly Mesoporous Carbon Derived from Cigarette Butt as Efficient Electro- catalysts for Zinc-Air Batteries, S. Hosseini, Chulalongkorn University
P-04	Wetting behavior of electrolyte in a GDE and how it affects performance in Zn-Air batteries, A. Kube, German Aerospace Center
P-05	Caffeine modified active carbon as catalyst layer in gas-diffusion electrodes for aprotic metal-air systems, I. Popov, Bulgarian Academy of Sciences
P-06	Performance Improvement of Lead Acid Batteries by Crystallized Graphite Powders in Negative Electrodes, H. Yang, National Chung Hsing University

Poster Program

	PEM FC + Electrolyser
P-07	Towards realistic benchmarking of fuel cell catalysts – Bridging the gap between RDE and MEA, A. Jensen, University of Copenhagen
P-08	In situ estimation of the effective (Nafion [®] & Anion) membrane Water diffu- sion coefficient in a PEMFC, K. Chadha, Pprime, University of Poitiers
P-09	Permeability for gaseous reactants – a key characteristic of PEM FC gas diffu- sion layer, M. Drakselová, University of Chemistry and Technology, Prague
P-10	Tuning the structure of Pt-Co catalyst materials for the oxygen reduction reaction in PEM fuel cell applications M. Singer, Carl von Ossietzky University of Oldenburg
P-11	Micro-CT imaging of electrolyser gas diffusion layers to extract pore space information for pore network modeling, H. Altaf, Fakultät für Verfahrens-und Systemtechnik, Otto von Guericke University Magdeburg
P-12	Simple and straight-forward thermal preparation of metal oxide nanoca- talysts from commercially available metal precursors for electrochemical water oxidation, A. Aziz, Center of Research Excellence in Nanotechno- logy (CENT), King Fahd University of Petroleum and Minerals
P-13	Catalytic active Manganese Oxide on Hydrophobic Nickel-Mesh for Appli- cation as Gas Diffusion Electrodes, A. Bekisch, Fraunhofer Institute for Ceramic Technologies and Systems IKTS
	CO ₂ Reduction
P-14	Electrochemical Reduction of CO ₂ on Copper Phosphide Catalysts in a Near Zero-gap Electrolytic Cell M. Choi, Gwangju Institute of Science Technology
P-15	Gas-Diffusion Electrodes for Practical CO ₂ Reduction: Challenges & Strategies G. El-Nagar, Helmholtz Zentrum Berlin für Materialien und Energie GmbH
P-16	High-entropy alloys as novel electrocatalysts for the CO ₂ reduction reaction J. Junqueira, Ruhr-Universität Bochum

Poster Program

P-17	Solvent effect on the ink preparation of Copper-based gas diffusion electro- des: A product distribution evaluation on the electrochemical reduction of CO ₂ , L. De Sousa, University Of Twente
P-18	Silver and Copper Nanoparticle Modified Gas Diffusion Electrodes for the CO ₂ Reduction Reaction. Impact of CO ₂ Back Pressure J. Weidner, Ruhr-Universität Bochum
	Processes In Alkaline Enviroment
P-19	Frequency Response Analysis of Oxygen Reduction Reaction in Concentra- ted Alkaline Solutions, S. Kandaswamy, Max Planck Institute Magdeburg
P-20	Workfunction of the Metal as Descriptor for Active and Durable Carbon Nanofiber Support Alkaline Oxygen Reduction Electrocatalyst S. Kang, Gwangju Institute of Science and Technology
P-21	Design of an In-Operando Cell for X-Ray and Neutron Imaging of Oxygen- Depolarized Cathodes in Chlor-Alkali Electrolysis M. Gebhard, Universität Bayreuth
P-22	Characterization of Bi-functional Oxygen Electrocatalysts K. Petrov, Bulgarian Academy of Sciences
P-23	NiCoPx Catalyst Based Electrode for Reversible Alkaline Fuel Cell J. Hnat, University of Chemistry and Technology Prague
P-24	Production of Bifunctional GDE for alkaline solutions HJ. Kohnke, Gaskatel GmbH
P-25	Novel alkaline water electrolysis process with nickel-iron gas diffusion elec- trodes for oxygen evolution, T. Turek, TU Clausthal

Poster Program

Notes

P-26One-Step Preparation of Ruthenium-CNT-Microtubes as Gas Diffusion Electrode for Electrochemical Ammonia Synthesis X. Wei, RWTH Aachen UniversityP-27Minimizing the influence of the electrolyte hydrostatic pressure on the potential distribution of hydrogen peroxide generating gas diffusion electrodes through electrode design and process parameter T. Muddemann, TU ClausthalP-28The expanded Micro Flow Cell for the side by side production and electro- chemical detection of Hydrogen Peroxide, J. Hübner, TU BerlinP-29Systematic building of an electroactive material library via gas-diffusion driven electrosynthesis, R. Prato, VITO NV / KU LeuvenP-30Application of gas diffusion electrodes in an electrochemical advanced oxidation process treating different synthetic wastewaters R. Simon, DECHEMA-ForschungsinstitutP-31Corrosion Current Density Related to Gas Diffusion Electrodes Durability Study, H. Yang, National Chung Hsing UniversityP-32Microbial reversible fuel cell integrated in membrane electrode assemblies G. Borisov, Bulgarian Academy of SciencesP-34Performance of gas-diffusion electrode as cathode for hydrogen generati- on in microbial electrolysis cell M. Mitov, South-West University, Blagoevgrad		Other Processes
P-27Minimizing the influence of the electrolyte hydrostatic pressure on the potential distribution of hydrogen peroxide generating gas diffusion electrodes through electrode design and process parameter T. Muddemann, TU ClausthalP-28The expanded Micro Flow Cell for the side by side production and electrochemical detection of Hydrogen Peroxide, J. Hübner, TU BerlinP-29Systematic building of an electroactive material library via gas-diffusion driven electrosynthesis, R. Prato, VITO NV / KU LeuvenP-30Application of gas diffusion electrodes in an electrochemical advanced oxidation process treating different synthetic wastewaters R. Simon, DECHEMA-ForschungsinstitutP-31Corrosion Current Density Related to Gas Diffusion Electrodes Durability Study, H. Yang, National Chung Hsing UniversityP-32Microbial ElectrobiotechnologyP-33Fabrication of bioanode for use in microbial electrolysis cell Y. Hubenova, Bulgarian Academy of SciencesP-34Performance of gas-diffusion electrode as cathode for hydrogen generati- on in microbial electrolysis cell M. Mitov, South-West University, Blagoevgrad	P-26	One-Step Preparation of Ruthenium-CNT-Microtubes as Gas Diffusion Elect- rode for Electrochemical Ammonia Synthesis X. Wei, RWTH Aachen University
P-28The expanded Micro Flow Cell for the side by side production and electro- chemical detection of Hydrogen Peroxide, J. Hübner, TU BerlinP-29Systematic building of an electroactive material library via gas-diffusion driven electrosynthesis, R. Prato, VITO NV / KU LeuvenP-30Application of gas diffusion electrodes in an electrochemical advanced oxidation process treating different synthetic wastewaters R. Simon, DECHEMA-ForschungsinstitutP-31Corrosion Current Density Related to Gas Diffusion Electrodes Durability Study, H. Yang, National Chung Hsing UniversityP-32Microbial ElectrobiotechnologyP-33Fabrication of bioanode for use in microbial electrolysis cell Y. Hubenova, Bulgarian Academy of SciencesP-34Performance of gas-diffusion electrode as cathode for hydrogen generati- 	P-27	Minimizing the influence of the electrolyte hydrostatic pressure on the po- tential distribution of hydrogen peroxide generating gas diffusion electro- des through electrode design and process parameter T. Muddemann, TU Clausthal
P-29Systematic building of an electroactive material library via gas-diffusion driven electrosynthesis, R. Prato, VITO NV / KU LeuvenP-30Application of gas diffusion electrodes in an electrochemical advanced oxidation process treating different synthetic wastewaters R. Simon, DECHEMA-ForschungsinstitutP-31Corrosion Current Density Related to Gas Diffusion Electrodes Durability Study, H. Yang, National Chung Hsing UniversityP-31Microbial ElectrobiotechnologyP-32Microbial reversible fuel cell integrated in membrane electrode assemblies G. Borisov, Bulgarian Academy of SciencesP-33Fabrication of bioanode for use in microbial electrolysis cell Y. Hubenova, Bulgarian Academy of SciencesP-34Performance of gas-diffusion electrode as cathode for hydrogen generati- on in microbial electrolysis cell M. Mitov, South-West University, Blagoevgrad	P-28	The expanded Micro Flow Cell for the side by side production and electro- chemical detection of Hydrogen Peroxide, J. Hübner, TU Berlin
P-30Application of gas diffusion electrodes in an electrochemical advanced oxidation process treating different synthetic wastewaters R. Simon, DECHEMA-Forschungsinstitut P-31Corrosion Current Density Related to Gas Diffusion Electrodes Durability Study, H. Yang, National Chung Hsing University P-31 Microbial Electrobiotechnology P-32Microbial reversible fuel cell integrated in membrane electrode assemblies 	P-29	Systematic building of an electroactive material library via gas-diffusion driven electrosynthesis, R. Prato, VITO NV / KU Leuven
P-31Corrosion Current Density Related to Gas Diffusion Electrodes Durability Study, H. Yang, National Chung Hsing UniversityMicrobial ElectrobiotechnologyP-32Microbial reversible fuel cell integrated in membrane electrode assemblies G. Borisov, Bulgarian Academy of SciencesP-33Fabrication of bioanode for use in microbial electrolysis cell Y. Hubenova, Bulgarian Academy of SciencesP-34Performance of gas-diffusion electrode as cathode for hydrogen generati- on in microbial electrolysis cell M. Mitov, South-West University, Blagoevgrad	P-30	Application of gas diffusion electrodes in an electrochemical advanced oxidation process treating different synthetic wastewaters R. Simon, DECHEMA-Forschungsinstitut
Microbial ElectrobiotechnologyP-32Microbial reversible fuel cell integrated in membrane electrode assemblies G. Borisov, Bulgarian Academy of SciencesP-33Fabrication of bioanode for use in microbial electrolysis cell Y. Hubenova, Bulgarian Academy of SciencesP-34Performance of gas-diffusion electrode as cathode for hydrogen generati- on in microbial electrolysis cell M. Mitov, South-West University, Blagoevgrad	P-31	Corrosion Current Density Related to Gas Diffusion Electrodes Durability Study, H. Yang, National Chung Hsing University
P-32Microbial reversible fuel cell integrated in membrane electrode assemblies G. Borisov, Bulgarian Academy of SciencesP-33Fabrication of bioanode for use in microbial electrolysis cell Y. Hubenova, Bulgarian Academy of SciencesP-34Performance of gas-diffusion electrode as cathode for hydrogen generati- on in microbial electrolysis cell 		Microbial Electrobiotechnology
P-33Fabrication of bioanode for use in microbial electrolysis cell Y. Hubenova, Bulgarian Academy of SciencesP-34Performance of gas-diffusion electrode as cathode for hydrogen generati- on in microbial electrolysis cell M. Mitov, South-West University, Blagoevgrad	P-32	Microbial reversible fuel cell integrated in membrane electrode assemblies G. Borisov, Bulgarian Academy of Sciences
 P-34 Performance of gas-diffusion electrode as cathode for hydrogen generation in microbial electrolysis cell M. Mitov, South-West University, Blagoevgrad 	P-33	Fabrication of bioanode for use in microbial electrolysis cell Y. Hubenova, Bulgarian Academy of Sciences
	P-34	Performance of gas-diffusion electrode as cathode for hydrogen generati- on in microbial electrolysis cell

